

Max. Marks: 90 Time: 90 Mins.

Entrance Test for Enrollment in Ph.D. Programme

Important Instructions

- Fill all the information in various columns, in capital letters, with blue/black ball point pen.
- > Use of calculators is not allowed. Use Blue/Black ball point pen for attempting the questions.
- > All questions are compulsory. No negative marking for wrong answers.
- To attempt a question, make a tick mark ($\sqrt{}$) at the right option/answer.
- Each question has only one right answer.
- > Questions attempted with two or more options/answers will not be evaluated.

Father's Name Roll No. Date: 15-01-2011 Signature of Candidate Signature of Invigilator Q.1 Hepatitis A and E are spread by: a) Skin contact b) Exhaled infected breath c) Blood transfusion and sexual intercourse d) Infected food and water c) Blood transfusion and sexual intercourse d) Infected food and water c) Inhibition of cell wall synthesis d) Inhibition of protein synthesis d) Inhibition of DNA synthesis lhibition of DNA synthesis lhibition of DNA synthesis Q.2 The correct equivalent for -10 °C is: a) -10 °F b) -18 °F c) 22 °F d) 14 °F c) 22 °F d) 14 °F e) Physostigmine and physostigmine lyhysostigmine and physostigmine lyhysostigmine and physostymine c) Quinidine and rescrpine ly Nicotine and neostigmine lyhysostymine and physostymine lyhysostymine and	Stream (Engg./Arch./Pharm./Mgmt./App.Sci./Life Sci.) Discipline		Pharmaceutical Sciences Pharmacy	
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		one would be the most irritating to the eye	; ·	only through an injury is:
a) Purified water a) Salmonella typhi	a)	Purified water	a)	Salmonella typhi
b) 0.9% NaCl solution b) Staphylococcus aureus	b)	0.9% NaCl solution		· -
c) 0.7% NaCl solution c) Clostridium tetani	c)	0.7% NaCl solution		
	d)	1.0% NaCl solution		

- Q.11 Which one of the following is not a calcium channel blocker:
- a) Nifedipine
- b) Ramipril
- c) Bepridil
- d) Diltiazem
- Q.12 Analysis of variance (ANOVA) is a statistical technique that can be used to find:
- a) Equality of means among group variables
- b) Substitution for random assignment
- c) Control of sequencing effects
- d) Equality of variances among group variances
- Q.13 Which is NOT true for the substance with the following chemical structure:

$$_{\mathrm{C}}$$
 $_{\mathrm{CH_{2}}}$ $_{\mathrm{CH_{2}}}$ $_{\mathrm{CH_{3}}}$

- a) Insoluble in water
- b) Insoluble in mineral acids
- c) Used as a topical local anesthetic
- d) Readily hydrolyzed in boiling water
- Q.14 Which of the following test can be employed to evaluate anxiety in experimental animals?
- a) Morris water maze test
- b) Elevated plus maze paradigm
- c) Rota rod test
- d) Tail flick test
- Q.15 One thousand pictograms are equal to:
- a) One femptogram
- b) One attogram
- c) One nanogram
- d) One microgram
- Q.16 A new drug is absorbed by transporters (i.e., carriers). Which dosage regimen will ensure maximum bioavailable fraction:
- a) 0.25 g TID
- b) 0.75 g OD
- c) 0.375 g BID
- d) All of the above
- Q.17 Insulin can't be absorbed orally, as it is:
- a) A macromolecule
- b) Poorly permeable
- c) Susceptible to gastric proteolytic enzymes
- d) All of the above
- Q.18 Gastroretentive systems are:
- a) Hydrodynamically balanced systems
- b) High density systems
- c) Bioadhesive systems
- d) Any of the above

- Q.19 Maximum number of compartments permitted in the non-compartmental, mamillary and physiological compartmental modeling are:
- a) 0, 3 and 10
- b) 0, 2 and 0
- c) 0, 2 and 20
- d) 0, 2 and 1
- Q.20 Which of the following statements is not true regarding nonparametric statistical tests:
- a) Used on small samples
- b) Can be applied to ordinal data
- c) Most popular choice is Student's t-test
- d) Can be used if data is non-normal
- Q.21 If a drug has a half life of 7 h, after how much time can the drug be assumed to have eliminated (> 99%) from the body:
- a) 7 h
- b) 49 h
- c) 70 h
- d) 14 h
- Q.22 Hepatic first-pass effect can not be avoided:
- a) By administering the drug sublingually
- b) By administering the drug intramuscularly
- c) By administering the drug per-orally
- d) By administering the drug subcutaneously
- Q.23 Ibuprofen has a pKa of 5.5. If the pH of the patient's urine is 7.5, what is the ratio of the dissociated to undissociated drug:
- a) 1:100
- b) 2:1
- c) 1:2
- d) 100:1
- Q.24 Loading dose of a drug is majorly based on:
- a) Time taken for complete elimination
- b) % of drug bound to plasma protein
- c) % of drug excreted unchanged in urine
- d) V_d & the desired plasma drug concentration
- Q.25 During competitive type of antagonism, the equilibrium constant (K_m) and total number of binding sites (V_{max}) get changed as:
- a) K_m is unchanged and V_{max} is reduced
- b) K_m is increased and V_{max} is unchanged
- c) K_m is increased and V_{max} is reduced
- d) None of the above
- Q.26 Characteristics of active transport of drugs include all of the following except:
- a) Selectivity by a specific carrier
- b) Energy dependence
- c) Possibility to occur against conc. gradient
- d) Non-saturability

0.27 The order of loss of sensory function when a Q.35 The characteristic yellow color of arachis oil local anaesthetic is used is: is due to the presence of: a) Temperature, pain, touch & deep pressure a) Sterols Touch, pain, deep pressure & temperature b) b) Carotenoids Pain, touch, deep pressure & temperature c) c) Flavonoids Pain, temperature, touch & deep pressure d) d) Squalene During application of various significance Q.28 0.36 The molecular weight of a drug is 602. tests, what is inferred by the term "p < 0.05": Approximately how many molecules would Probability of success in results is 5% or less a) be present in 1 mg of the drug: Probability to reject null hypothesis is < 5% b) 10^{18} a) Probability to accept null hypothesis is < 5% c) 10^{20} b) Probability of failure in results is 5% or less d) 10^{17} c) 10^{23} d) Q.29 Atherosclerosis and high blood pressure are related to each other on the basis of: Q.37 The renal excretion of a weakly acidic drug a) Charles's law of pKa 3 will be more rapid in alkaline urine b) Boyle's law than in acidic urine because: c) Hagen-Poiseuille equation a) All drugs are excreted rapidly in basic urine Newton's law of flow d) b) Drug will be unionized; not easily reabsorbed Drug will be ionized; not easily reabsorbed c) Quinolones should not be administered with: Q.30 d) Active transport functions better in basic urine a) Dairy products Q.38 Which of the following statements is not true Antacids b) about warfarin: c) Vitamin-Mineral combinations d) All of the above a) Has higher molecular weight than heparin b) Acts within 24 hours of administration Q.31 Volume of distribution for a drug is: c) Is metabolized chiefly by liver d) NSAIDs cause hemorrhage with warfarin a) Sum of intracellular & extracellular volumes Total blood volume b) Q.39 Official Indian drug regulatory agency is: c) Ratio of drug dose to plasma drug levels Ratio of body weight to body density d) a) PCI. India b) DCGI, India Clinically available anticancer drugs have 0.32 c) AICTE, India one of the following underlying mechanism d) FDA, India of action: Improving body defense mechanism Q.40 Level A IVIVC is established between: a) Receptor site blockade of cancer cell b) a) MDT and MRT Inhibition of cell wall synthesis c) b) T50% and Cmax Cell growth inhibition d) c) Percent dissolved and percent absorbed d) T_{60%} and AUC The absorption of which of the following Q.33 NSAIDs is not affected when antacids are Q.41 "Ganja" as per the Narcotic Drugs and administered concurrently: Psychotropic Substances Act means a constituent onbtained from Cannabis sativa, a) Fenoprofen b) Sulindac consisting of its: Piroxicam c) a) Flowering and fruiting tops d) None of the above b) c) Leaves, seeds, flowering tops & fruiting tops Q.34 Which of the following is not bactericidal in d) All parts of the plant its action: Q.42 Penicillins a) Good and bad lipoproteins, respectively, are: b) Tetracyclines a) LDL and cholesterol Cephalosporins c) b) LDL and VLDL

c)

d)

HDL and VLDL

VLDL and cholesterol

d)

Isoniazid

Punarnava consists of the whole herb of 0.51 Most of the skin diseases are treated using: Boerhaavia diffusa and it is used in: a) Ocimum sanctum a) Gonorrhea Terminalia chebula b) b) Cardiac diseases c) Azadirachta indica Liver diseases d) Saraca indica c) d) Cold and cough O.52 Which of the following correlations is the In severe alcohol intoxication, the blood Q.44 strongest: ethanol levels exceed: -0.9997 a) 150 ng/dL -0.0001 a) b) 300 ng/dLb) c) +0.9031250 ng/dL d) +0.5000c) 400 ng/dL d) Usually, the volatile oils are the mixture of: Q.53 0.45 DNA fingerprinting does not help in: a) Mono and sesquiterpenes Solving paternity disputes Sesquiterpenes and diterpenes a) b) Curing genetic diseases Mono and diterpenes b) c) Monoterpenes & aliphatic compounds c) Identifying criminals in sexual crimes d) d) Authenticating plant breeding programmes Q.54 Which of the following factors is also known Q.46 Cetyl trimethyl as the lipid solubility of an anaesthetic agent: ammonium bromide (Cetrimide, Savlon) is not administered Blood: gas partition coefficient a) orally, as it is: b) Minimum alveolar concentration Poorly permeable a) c) Oil: gas partition coefficient Poorly soluble d) Oil: blood partition coefficient b) Corrosive c) Balsams are resinous substances which d) Highly toxic Q.55 contain large proportions of: A drug used to "kill" viruses is: Q.47 a) Resins and volatile oils a) Idoxuridine Glycosides and resin b) Acyclovir b) Gum, resin and volatile oil c) c) **AZT** d) Benzoic acid, cinnamic acid & their esters d) None The antidote for cyanide poisoning is: Q.56 Q.48 Radioactive iodine is used in treatment of: Dimercaprol a) a) Hypothyroidism b) Hydroxocobalamin **EDTA** b) Hyperthyroidism c) Iodinemia d) No antidote c) d) Goitre Q.57 Chemically, ibuprofen is: In chronic alcoholics and chain smokers, the 0.49 a) 2-(4-Isobutylphenyl) propionic acid drug dose has to be invariably: Phthalazin-1-yl hydrazine b) Increased 3, 20-Dioxo-4-pregen-17α-yl benzoate a) c) Decreased 2,2-Dihydroxy-hexachloro diphenyl methane b) d) Unchanged c) d) Stopped Q.58 Sarson oil and other vegetables contaminated with argemone can cause: Q.50 Which of the following solvents is not used Glaucoma a) in NMR spectroscopy: Dropsy b) CCl_4 a) c) **Paralysis** b) CHCl₃ d) Liver cirrhosis CS_2 c) CF₃COOH

d)

0.59 Indicate the correct order of increasing eluent 0.67 All of the following drugs can be assayed by complexometric titration except: power of the following solvents: a) Chloroform, benzene, ethyl acetate, ether a) Alumina gel b) Ether, chloroform, ethyl acetate, benzene b) Calcium lactate Benzene, ether, chloroform, ethyl acetate Magnesium sulphate c) c) Ethyl acetate, ether, benzene, chloroform Acetyl salicylic acid d) d) The antiarrythmic drug, quinidine, is a: In reversed phase HPLC, the mobile and Q.60 Q.68 stationary phase, respectively, are: a) (+) Stereoisomer of quinine Racemic mixture of quinine b) a) Polar and nonpolar (-) Stereoisomer of quinine c) b) Both are nonpolar d) None of the above Nonpolar and polar c) Both are polar d) Q.61 Infected blood products may produce serum hepatitis due to presence of: 0.69 Which of the following statements is incorrect regarding the data "4, 6, 9, 4, 2": a) Hepatitis A virus Hepatitis C virus Mean of the data is 5 b) a) c) Hepatitis B virus b) Mode of the data is 4 None of the above Median of the data is 9 d) c) Range of the data is 7 d) Q.62 Which of the following is not an ethical guideline for conducting research with Q.70 Rancidity of fats is due to: human participants? Oxidation a) a) Getting informed consent of the participant b) Hydrolysis b) Keeping participants' identity anonymous Saponification c) c) Telling them to continue till study completes d) Neutralization Telling "you are free to withdraw any time" d) 0.71 All of the following diuretics inhibit Alkyl substitution in an alkene causes the carbonic anhydrase, except: Q.63 following spectroscopic effect: a) Acetazolamide Hypochromic Furosemide a) b) b) Bathochromic c) Bumetanide Hyperchromic c) d) Ethacrynic acid Hypsochromic d) Q.72 Substances which are weakly acidic or basic In a double beam spectrophotometer, the yield sharp end point if titrated with: Q.64 primary source of light is divided into two Strong acid a) beams of: Weak base b) a) Standard intensity c) Strong base Equal intensity b) d) Non-aqueous solvent Standard and sample intensity c) d) Visible light intensity Q.73 The functional group which contributes to the instability of aspirin is: Q.65 A fatty acid which is not synthesized in Alcohol a) human body and has to be supplied in diet is: Ester b) Palmitic acid a) c) Ketone b) Linoleic acid d) Carboxylic acid Oleic acid c) d) Stearic acid Q.74 The main use of *Piper nigrum* is as: Aromatic stimulant a) Pyrolysis gas chromatography is useful for Q.66 b) Antipyretic the analysis of: Antiperiodic c) Halogenated compounds a) d) Antidiarrhoeal b) High molecular weight compounds

Amino acids

Low molecular weights compounds

c)

d)

An example of synthetic antimicrobial is: 0.75 Approximately how many mL of sulphuric 0.83 acid (specific gravity 1.3) are required to Benzyl penicillin a) prepare 1 L of 0.1 N acid? b) Cloxacillin sodium Amphomycin c) 7 a) Ciprofloxacin d) b) 50 70 c) Q.84 A drug is being metabolized by zero-order d) 5 kinetics. This indicates that a constant: 0.76 Viagra contains anti-impotency drug: a) Drug amount is catabolized every hour a) Viagrinil hydrochloride b) Drug amount is present in body at any time Drug fraction is catabolized every hour Nitric oxide c) b) Time is needed to catabolize half of the drug d) Sildenafil citrate c) d) Stimulate nitrate 0.85 Grant of a patent, under "The Patents Act of India, 1970" may be opposed by any person In the gel permeation chromatography, Q.77 from the date of advertisement of acceptance molecules are separated on the basis of: within a period of: a) Chemical nature 4 months a) Adsorptive properties b) 3 years b) Size and shape c) 10 months c) d) Partition coefficient d) 2 years O.78 If the chloroform extract of a drug on Q.86 Oral rehydration mixture contains glucose addition of ammonia gets changed to red and sodium, because both of them are: color, it indicates the presence of: Needed to maintain the plasma osmolality a) Cardiac glycosides a) Help in G.I. absorption of each other b) Amygdalin b) Are prominent energy sources for the body c) c) Sennosides Are required for activation of Na-K ATPase d) d) Saponins A research scholar needs to prepare 40% v/v Q.87 Q.79 Approximately, what percentage of score alcohol. In what proportion, he should mix falls within one (±1) standard deviation of 95% alcohol and water to obtain the desired the mean in a normal distribution? concentration of alcohol: 8:11 a) a) 34% b) 10:9 b) 99% 11:8 c) c) 95% 9:10 d) 68% d) If a drug has been prepared, packed and Q.88 Q.80 The propellant used for topical aerosols is: stored under insanitary conditions, where it may have been rendered injurious to health, a) Propane Trichloromonofluoro methane it may be termed as: b) c) Nitrous oxide Toxic drug a) Dichloro difluoro methane d) b) Misbranded drug Adulterated drug c) A fluid which requires a definite initial stress Q.81 d) Spurious drug to start flowing is termed as: Q.89 The average pH of urine is close to: a) Dialatant 5.0 a) b) Pseudoplastic 8.0 b) Plastic c) c) 6.0 Thixotropic d) d) 7.0 Q.82 Which species, having an erythrocytic cycle Q.90 Though rich in nutrients, milk is deficient in: of 48 h, is the common cause of malaria, the Vitamin A a) exo-erythrocytic form of which may remain Vitamin B2 b) for years, thus leading to relapses: Vitamin C c) a) Plasmodium falciparum d) Vitamin D b) Plasmodium ovale Plasmodium vivax c)

d)

Plasmodium malariae